



UNIVERSITAT DE
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Double carbon structure energy storage device



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Experimental Procedure



Working plan



Two major challenges
facing human
sustainable
development in the
21st century - **energy**
and **environmental**
issues



Electrochemical Energy storage device

Rocking chair cell

Battery

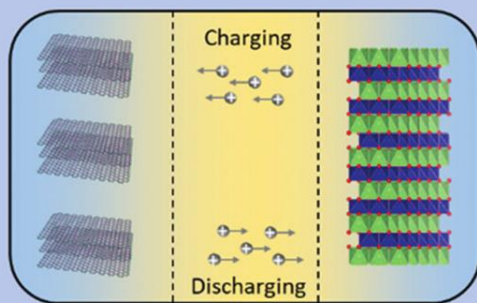
Electrochemical capacitor

Dual-ion cell

Hybrid capacitor

capacitor

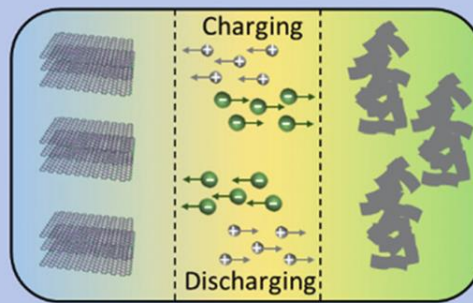
Battery



Battery-type anode

Battery-type cathode

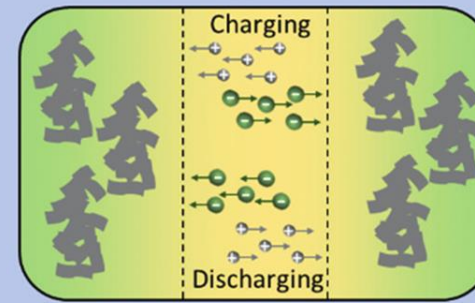
Hybrid Capacitor



Battery-type anode

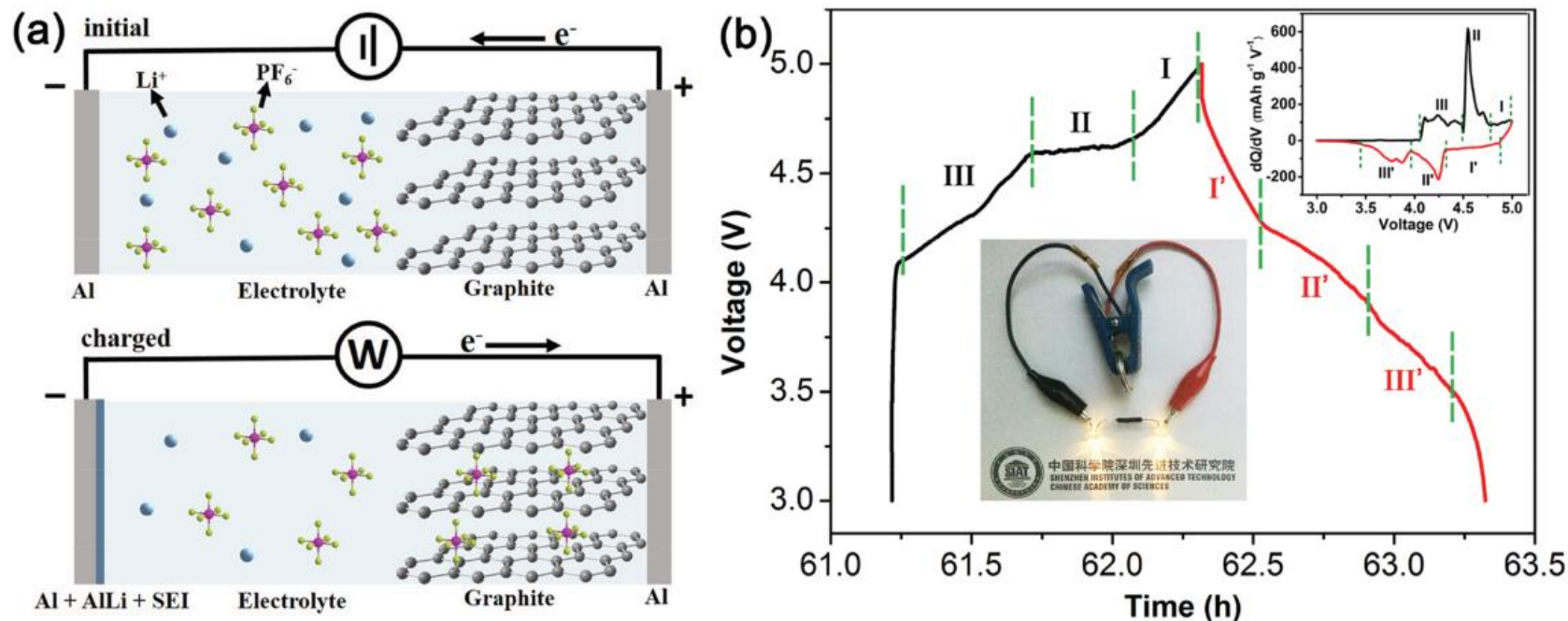
Capacitor-type cathode

Capacitor



Capacitor-type anode

Capacitor-type cathode

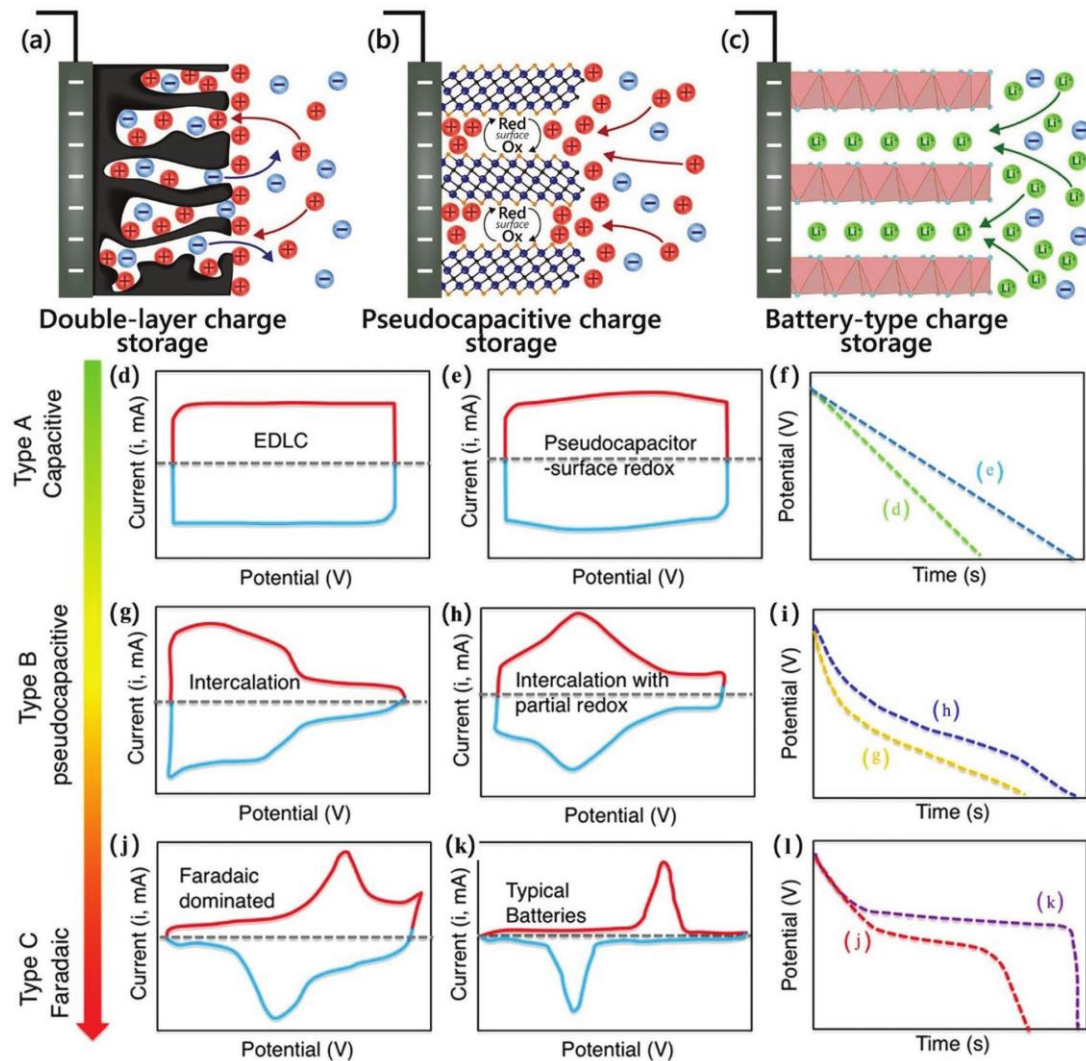


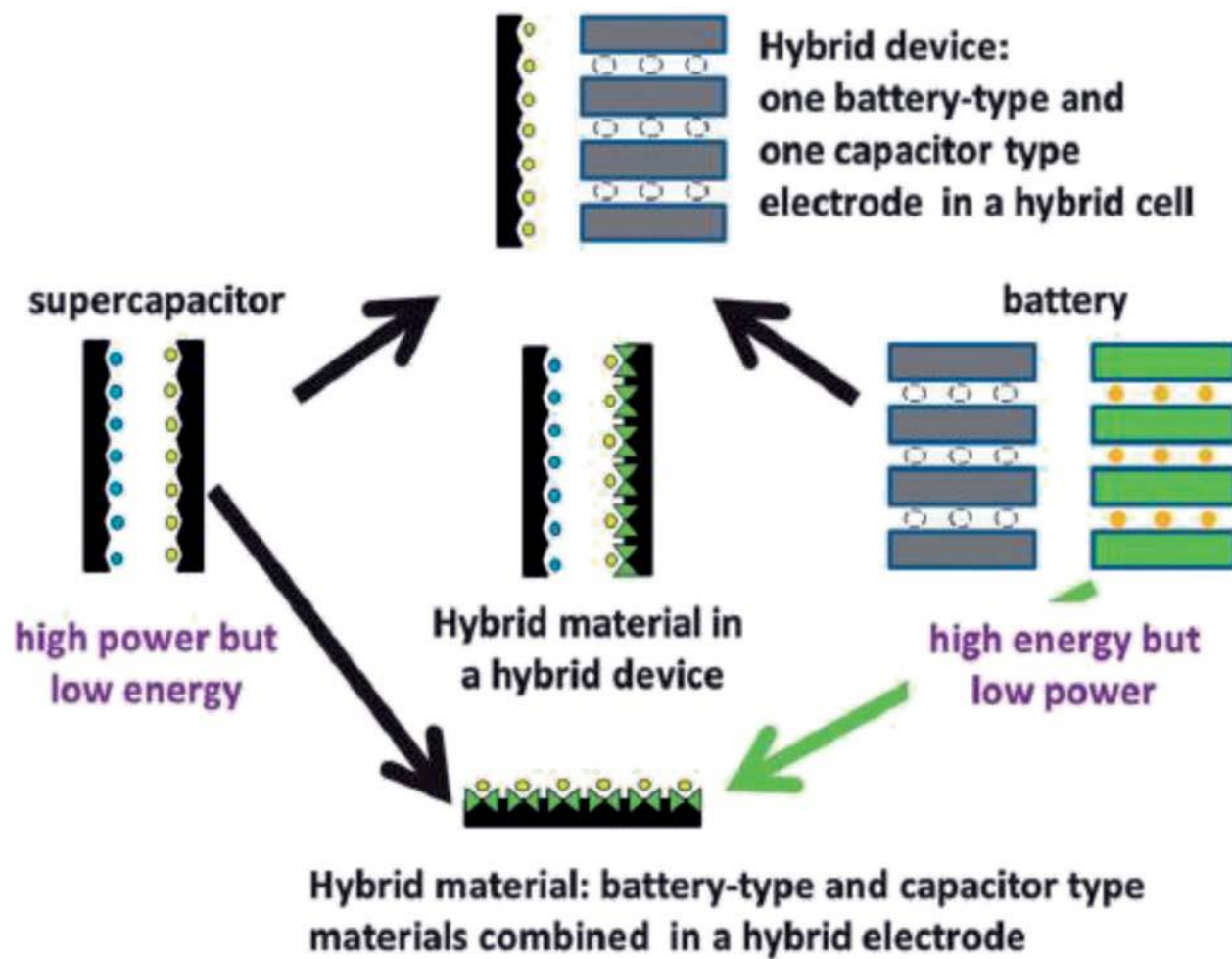
Anode: $\text{Al} + \text{Li}^+ + \text{e}^- \leftrightarrow \text{AlLi}$

Cathode: $x\text{C} + \text{PF}_6^- \leftrightarrow \text{C}_x(\text{PF}_6) + \text{e}^-$

Electrolyte: 4M LiPF₆ in EMC + 2 wt.% VC

Schematic illustration of the DIB in the initial state (up) and the charged state (below). b) Galvanostatic charge–discharge curve of the AGDIB at 0.5 C (1 C corresponding to a current rate of 100 mA g⁻¹). Insets are the dQ/dV differential curve of the battery and a photograph showing that a single AGDIB cell lighting up two yellow LEDs in series.

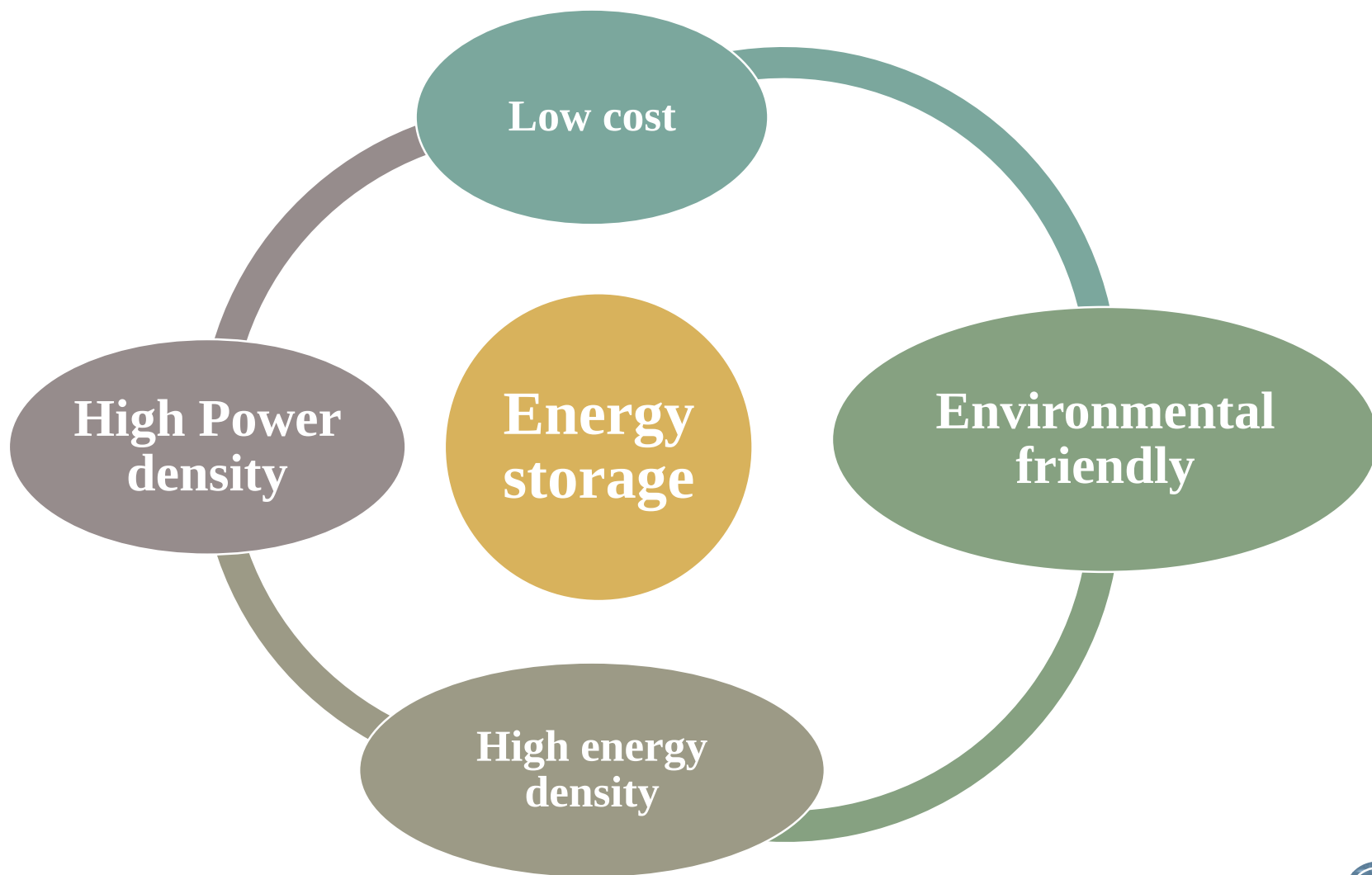


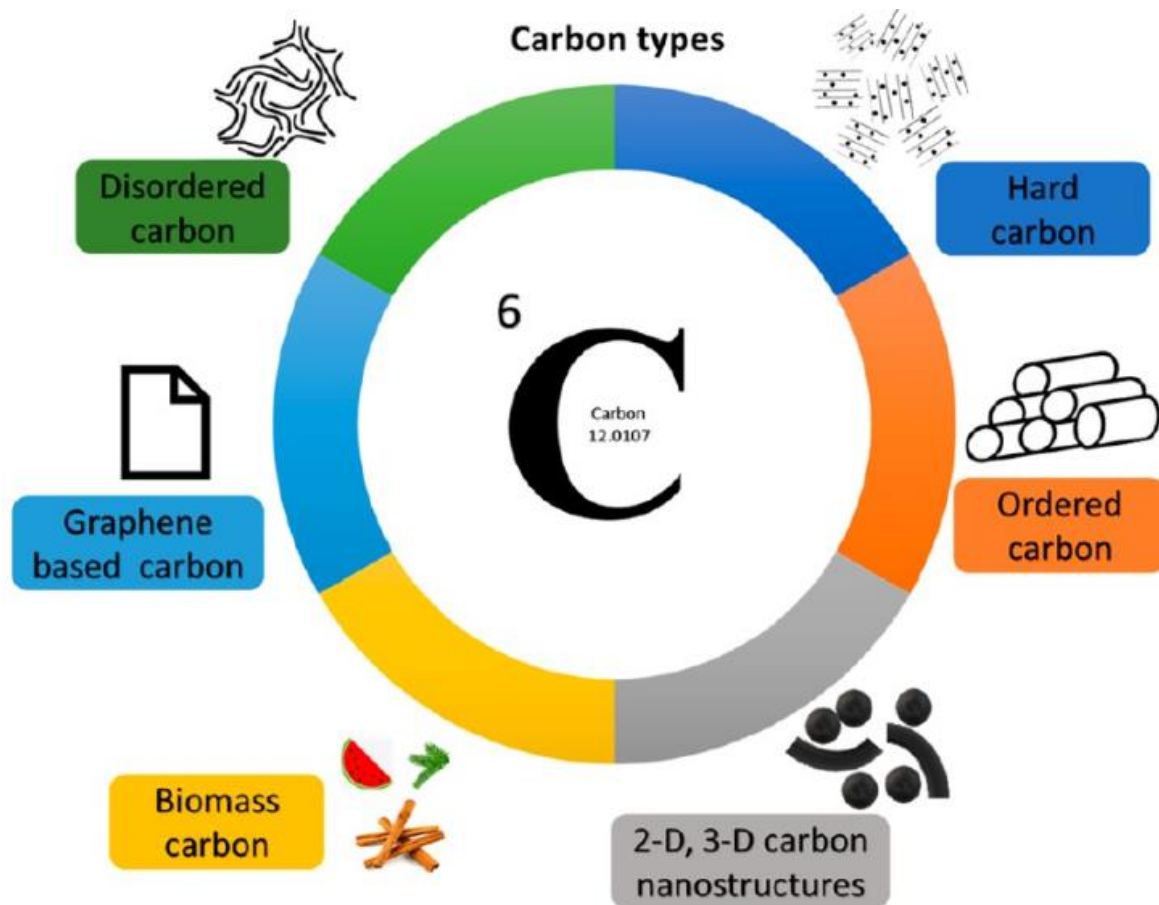




Motivation

Motivation and Objectives





Advantage:

- Low cost
- Abundant
- Environmentally friendly
- Conductivity
- Stability



Motivation

Motivation and Objectives

Supercapacitor

Advantage:

1. High power density
2. Long cycle
3. Low cost

Disadvantage:

1. Low energy density

Active carbon

Graphite

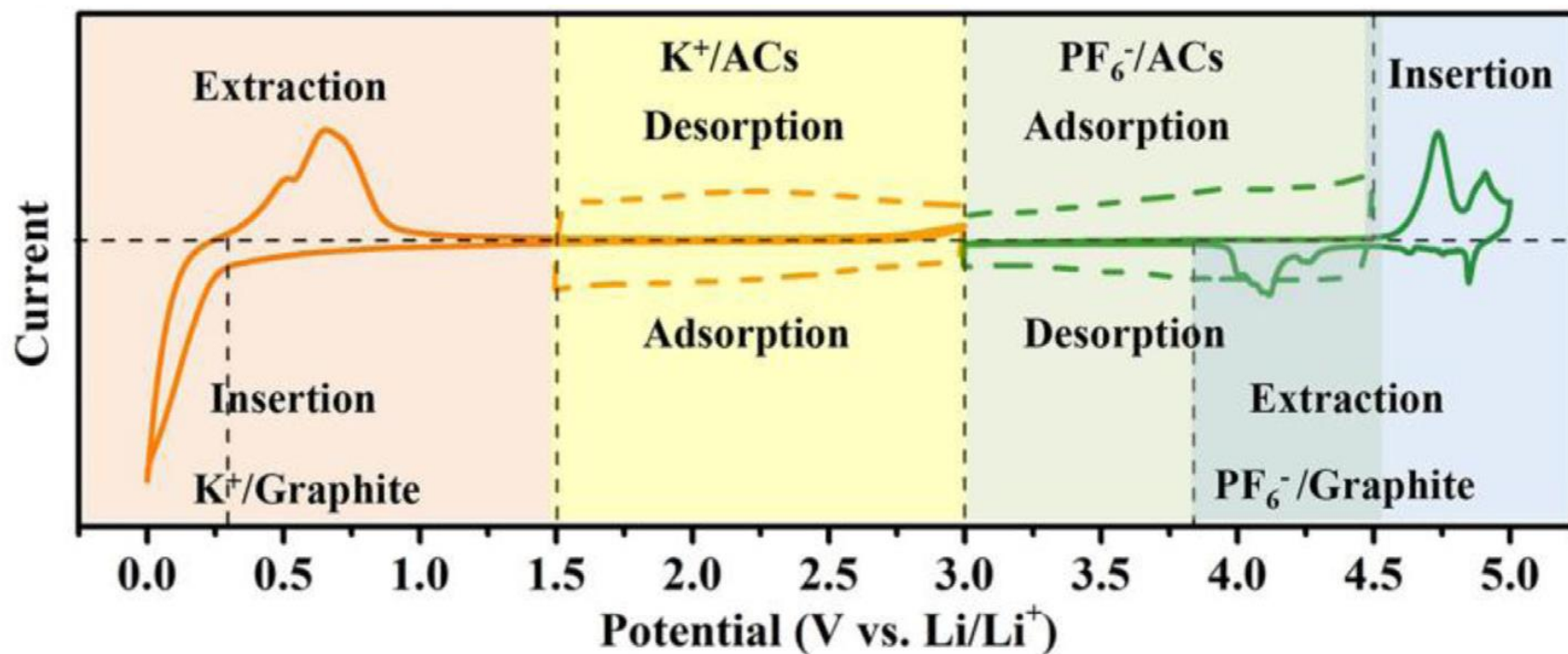
Dual-ion batteries

Advantage

1. High energy density
2. High voltage window
3. Low cost

Disadvantage:

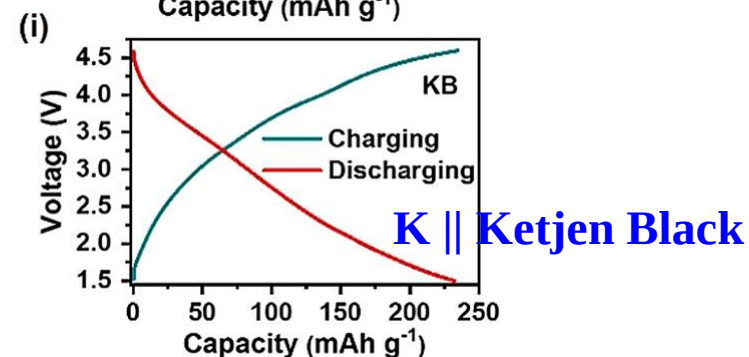
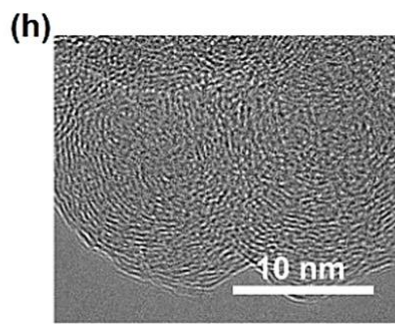
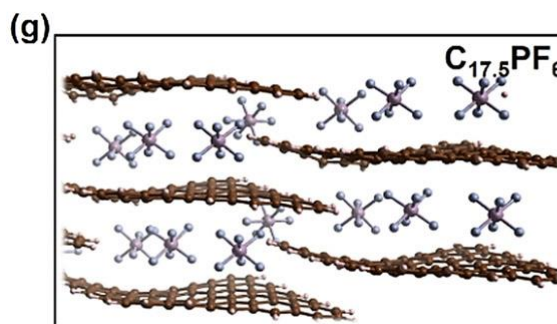
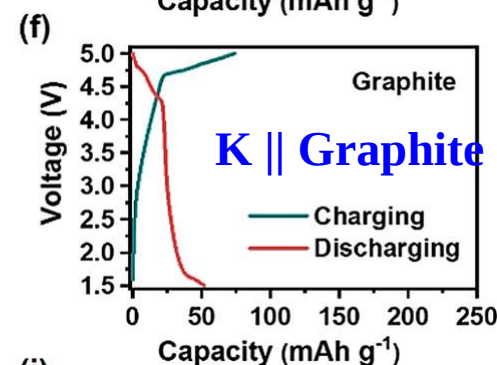
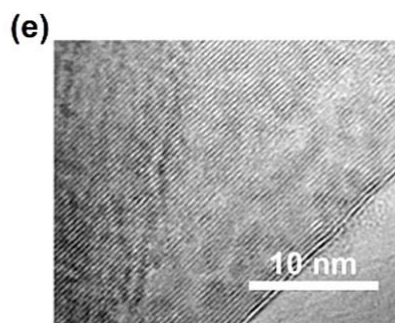
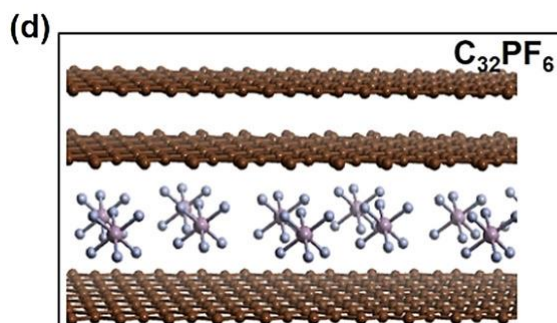
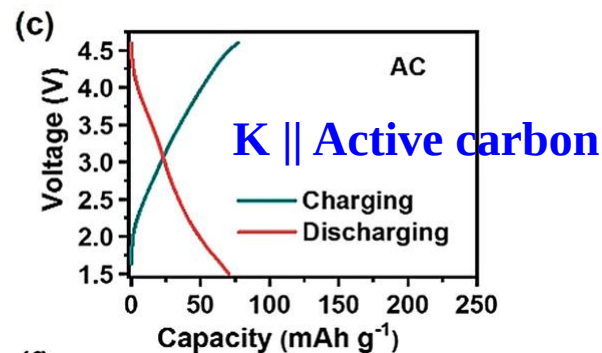
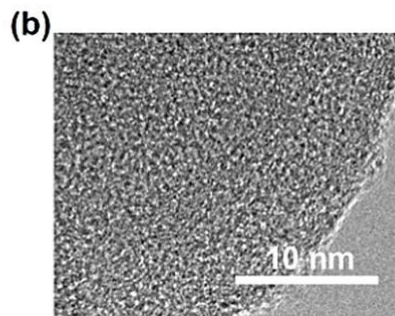
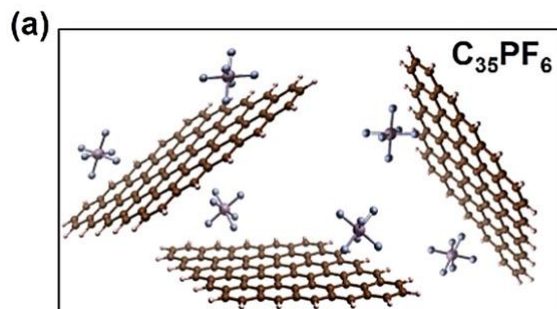
1. Low power density
2. Low cycle-number



The theoretical basis for building integrated devices. CV curve of K⁺-intercalation graphite anode at 0.5 mV/s at an operating potential window (OPW) of 0–3.0 V (orange solid line) and CV curve of PF₆⁻-intercalation graphite cathode at 0.5 mV s⁻¹ at an OPW of 3.0–5.0 V (olive solid line); CV curve of K⁺-absorption on AC at 5 mV/s at an OPW of 1.5–3.0 V (orange dotted line) and the CV curve of PF₆⁻-absorption on AC at 5 mV/s at an OPW of 3.0–4.5 V (olive dotted line). (c) Schematic of the electrochemical stable windows (SWs) and the electrochemical unstable windows (UWs) of 0.8 M KPF₆ in EC/DEC (1/1, v/v) based on activated carbon (AC) and graphite. The solid electrolyte interface (SEI) film forms in the region of 0–0.8 V versus Li/Li⁺.

Graphite Insertion/Adsorption

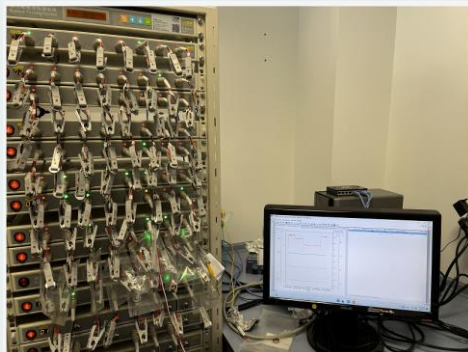
Motivation and Objectives



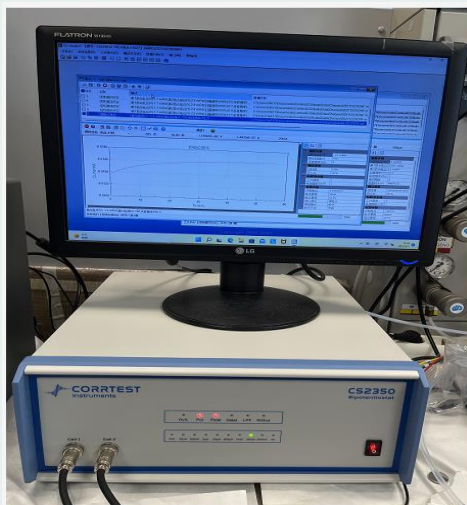


Characterization

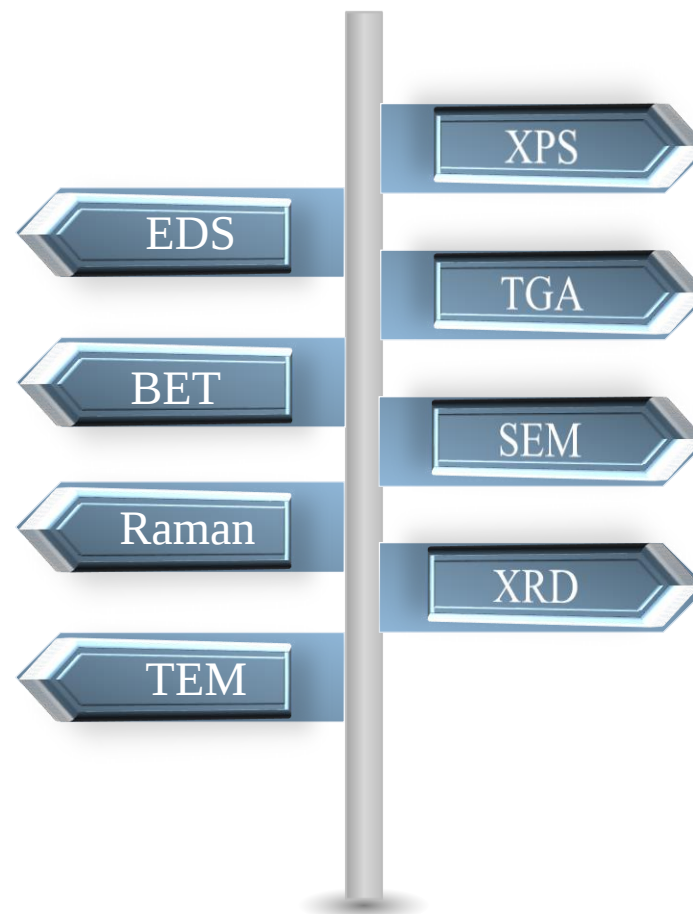
Experimental Procedure



Neware battery test system



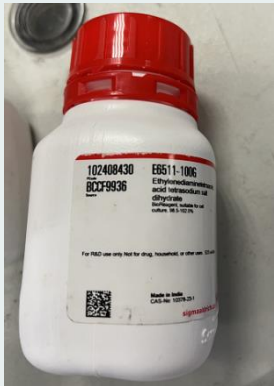
Correst electrochemical Workstation



Material Characterization Facilities.

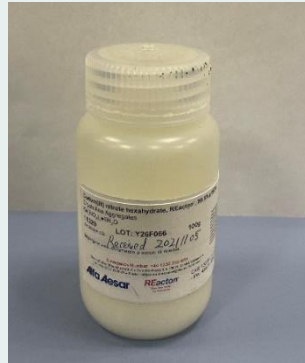
Raw materials

EDTA

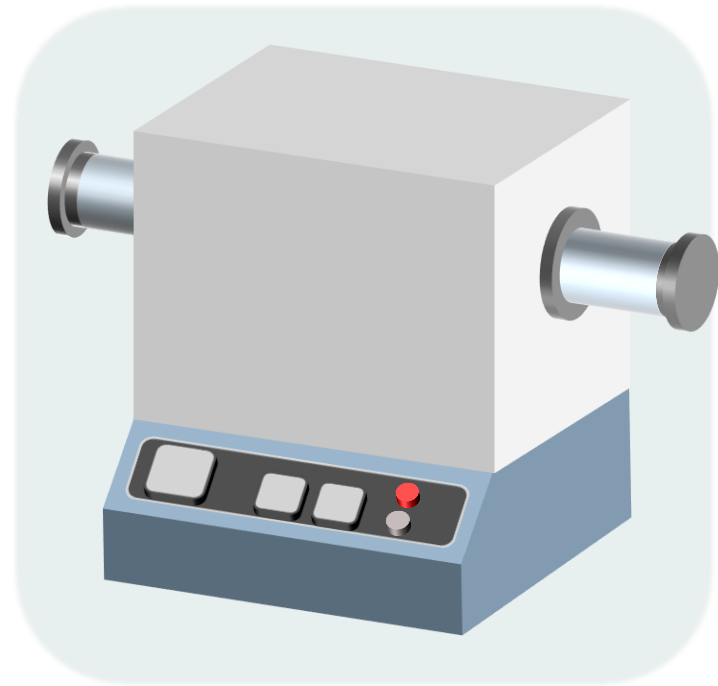


+

$\text{Co}(\text{NO}_3)_2$



Preparation method



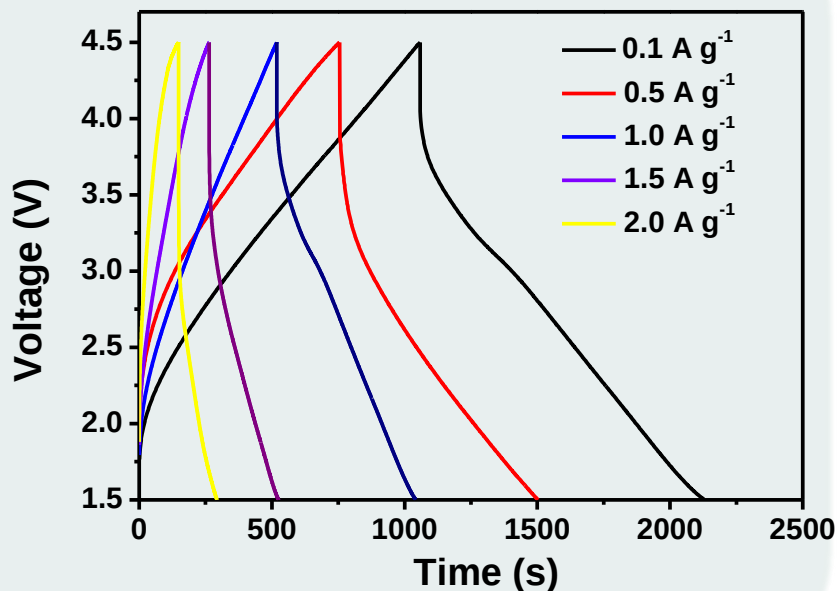
Cobalt Catalyzed Graphitization Carbon



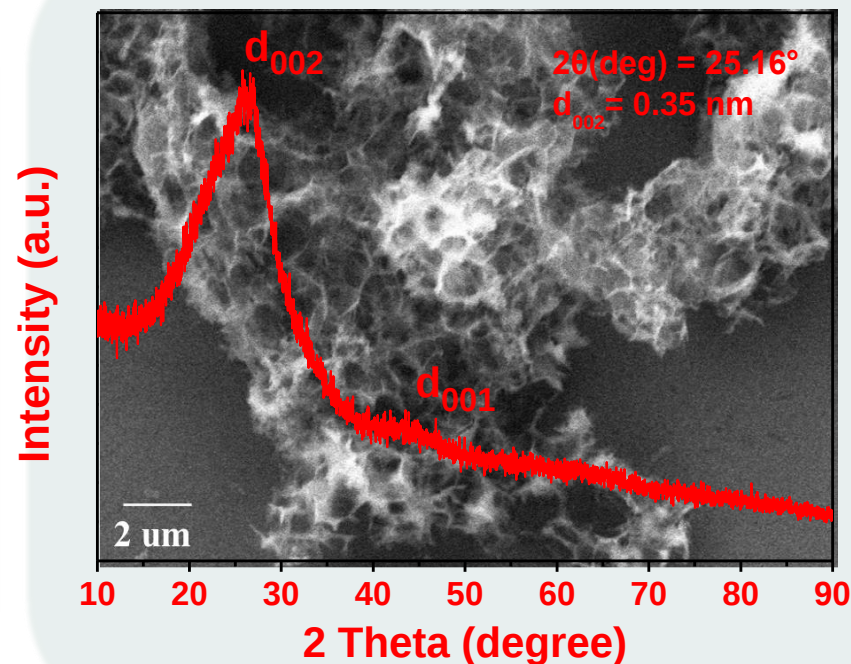
Preliminary work

Experimental Procedure

3D highly graphitized porous carbon



Hard carbon || 3D highly graphitized porous carbon



Next Plan:

1. Extended the double-carbon structure energy storage device;
2. Increase power density



This project is planned to take place from October 2021 until the end of April 2024.

The chronogram of the project is shown as follows:

Time schedule.

Literature review										
WP1. Initiating to design the experimental device										
WP2. Preparation of graphitization carbon anode material										
WP3. Preparation of graphitization carbon cathode material										
WP4. Design of electrolyte system suitable for DCESD system										
WP5. Development of new DCESD system.										
WP6. Summarize all data into papers.										

**Thanks for your
attention!**